

Qing-Xiang Liu,* Jin-Cheng Feng,
Li-Na Yin and Jian-Hua GuoCollege of Chemistry and Life Science, Tianjin
Normal University, Tianjin 300074, People's
Republic of China

Correspondence e-mail: qxliu@eyou.com

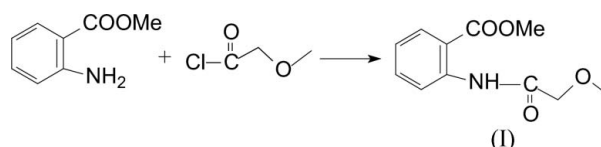
Key indicators

Single-crystal X-ray study
 $T = 293$ K
Mean $\sigma(\text{C}-\text{C}) = 0.004$ Å
 R factor = 0.057
 wR factor = 0.170
Data-to-parameter ratio = 13.2For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

Methyl 2-(methoxyacetylamino)benzoate

The title compound, $\text{C}_{11}\text{H}_{13}\text{NO}_4$, was obtained by reaction of
methyl 2-aminobenzoate and 2-methoxyacetyl chloride. The
two side chains lie in the same plane as the benzene ring.Received 5 July 2006
Accepted 7 July 2006

Comment

Methyl 2-(methoxyacetylamino)benzoate, (I), is an important
intermediate in the chemical industry (Aavid *et al.*, 1949) and
is used extensively in medicine and pesticides (Stephen *et al.*,
2002). For example, it is a precursor of quinazoline derivatives
and a potential anticancer medication (Silvana *et al.*, 2004). We
report here the synthesis and crystal structure of the title
compound, (I). The two side chains lie in the same plane as the
benzene ring plane (dihedral angle 1.4°).

Experimental

2-Methoxyacetyl chloride (5.38 g, 0.05 mol) was added dropwise over
a period of 30 min to a dichloromethane (60 ml) solution of methyl 2-
aminobenzoate (5.00 g, 0.033 mol) and pyridine (5.22 g, 0.066 mol) in
a cool water bath (273–278 K), then the mixture was stirred for 2 h at
room temperature. The resulting solution was washed with 1M HCl
(50 ml), 1 M NaOH (50 ml), saturated NaCl (50 ml) and water
(50 ml). The dichloromethane solution was dried over MgSO_4 , and
the solvent was removed using a rotary evaporator to give (I) as a
white solid (yield: 4.80 g, 65%; m.p. 345–346 K). Crystals of (I)
suitable for X-ray diffraction were obtained by slow evaporation of a
 CH_2Cl_2 solution at room temperature. Analysis calculated for
 $\text{C}_{11}\text{H}_{13}\text{NO}_4$: C 59.19, H 5.87, N 6.28, O 28.67%; found: C 59.33, H 5.54,
N 6.44, O 28.76%.

Crystal data

$\text{C}_{11}\text{H}_{13}\text{NO}_4$	$V = 555.8$ (5) Å ³
$M_r = 223.22$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.334$ Mg m ⁻³
$a = 7.149$ (4) Å	Mo $K\alpha$ radiation
$b = 8.649$ (5) Å	$\mu = 0.10$ mm ⁻¹
$c = 9.973$ (5) Å	$T = 293$ (2) K
$\alpha = 67.462$ (5)°	Prism, colorless
$\beta = 89.982$ (7)°	$0.24 \times 0.16 \times 0.12$ mm
$\gamma = 78.350$ (7)°	

Data collection

Bruker SMART CCD area-detector diffractometer	3000 measured reflections
φ and ω scans	1941 independent reflections
Absorption correction: multi-scan (SADABS; Shelldrick, 1996)	1166 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.975$, $T_{\max} = 0.988$	$R_{\text{int}} = 0.029$
	$\theta_{\text{max}} = 25.0^\circ$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.057$ $wR(F^2) = 0.170$ $S = 1.03$

1941 reflections

147 parameters

H-atom parameters constrained

 $w = 1/[\sigma^2(F_o^2) + (0.0937P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\max} = 0.001$ $\Delta\rho_{\max} = 0.21 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{\min} = -0.23 \text{ e } \text{\AA}^{-3}$

All H atoms were initially located in a difference Fourier map, but were then constrained to an ideal geometry, with C—H = 0.93–0.97 Å and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$, and refined as riding on their parent atoms.

Data collection: *SMART* (Bruker, 2000); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 2000); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997a); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997a); molecular graphics: *SHELXTL* (Sheldrick, 1997b); software used to prepare material for publication: *SHELXTL*.

This project was supported by Tianjin Normal University Personnel Division, No 5r1036.

References

- Aavid, T. Z. & Wagner, E. C. (1949). *J. Org. Chem.* **14**, 967–977.
 Bruker (2000). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
 Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.
 Sheldrick, G. M. (1997a). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.

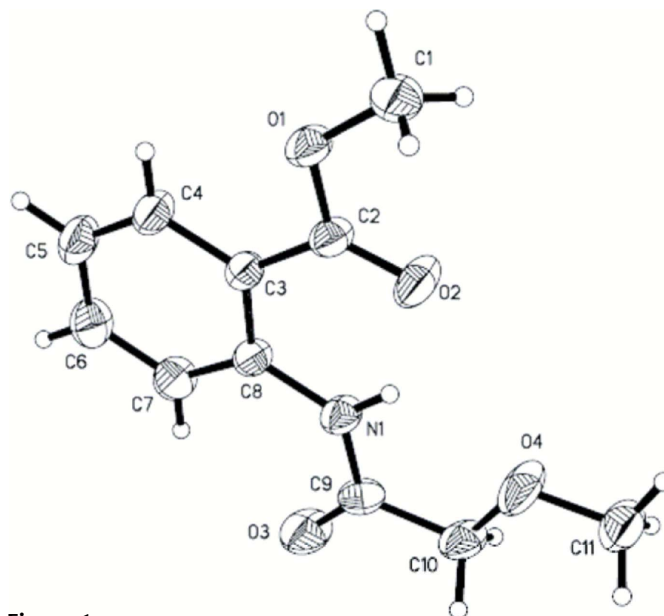


Figure 1

A view of (I) with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

- Sheldrick, G. M. (1997b). *SHELXTL*. Version 5.10 for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.
 Silvana, A. R., Jose, E. B. & Roberto, R. (2004). *Synthesis*, pp. 429–435.
 Stephen, W. W., Anthony, A. C. & Maynard, D. C. (2002). *J. Med. Chem.* **45**, 3865–3877.